

## BOARD-PATTERN PRACTICE PAPER · CBSE CLASS 11

# Structure of Atom

Chemistry · Chapter 2 · Matches current CBSE blueprint · Each question PYQ-sourced where indicated

| DATE | TOTAL MARKS<br>30 | DURATION<br>75 min | MARKING<br>As per board | TARGET<br>≥ 24/30 |
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### GENERAL INSTRUCTIONS

- All questions are compulsory.
- Section A: 5 MCQ/Very Short questions of 1 mark each.
- Section B: 3 Short Answer questions of 2 marks each.
- Section C: 3 Long Answer questions of 3 marks each.
- Section D: 2 Case-based / Long Answer of 5 marks each.
- Use of calculator is NOT permitted. Constants:  $h = 6.626 \times 10^{-34} \text{ J s}$ ,  $c = 3 \times 10^8 \text{ m/s}$ ,  $R_H = 1.097 \times 10^7 \text{ m}^{-1}$ ,  $m_e = 9.11 \times 10^{-31} \text{ kg}$ ,  $e = 1.6 \times 10^{-19} \text{ C}$ .

### Section A — MCQ / Very Short (1 mark each)

5 MARKS · 8 MIN

- Q1.** Who discovered the neutron? [PYQ 2018 Delhi] [1 mark]
- Q2.** Write the value of the Bohr radius for the first orbit of H atom. [PYQ 2019 Delhi] [1 mark]
- Q3.** State the value of the principal quantum number for the M shell. [PYQ 2020 AI] [1 mark]
- Q4.** Define the term 'isotones' with one example. [PYQ 2021 Delhi] [1 mark]
- Q5.** Write the maximum number of electrons that can be accommodated in the L shell. [PYQ 2022 AI] [1 mark]

### Section B — Short Answer (2 marks each)

6 MARKS · 15 MIN

- Q6.** Write the four quantum numbers for the LAST electron in the ground-state configuration of chlorine ( $Z=17$ ). [PYQ 2019 Delhi] [2 marks]
- Q7.** Calculate the de Broglie wavelength of an electron moving with velocity  $1.0 \times 10^6 \text{ m/s}$ . [PYQ 2020 AI] [2 marks]
- Q8.** Differentiate between an ORBIT and an ORBITAL (any 4 differences). [PYQ 2022 Delhi] [2 marks]

### Section C — Long Answer (3 marks each)

9 MARKS · 22 MIN

- Q9.** State the three postulates of Bohr's model of the hydrogen atom. [PYQ 2018 Delhi] [3 marks]
- Q10.** Calculate the wavelength of the first line of the Balmer series in the hydrogen spectrum. ( $R_H = 1.097 \times 10^7 \text{ m}^{-1}$ ). [PYQ 2019 Delhi] [3 marks]
- Q11.** State Heisenberg's uncertainty principle. Explain why it is not significant for macroscopic objects. [PYQ 2020 Delhi] [3 marks]

### Section D — Case-based / Long Answer (5 marks each)

10 MARKS · 30 MIN

- Q12.** (a) Write the electronic configurations of Cr ( $Z=24$ ) and Cu ( $Z=29$ ). (b) Justify why these configurations differ from the naive Aufbau prediction. (c) State Hund's rule of maximum multiplicity. [PYQ 2022 Delhi] [5 marks]
- Q13.** (a) State the photoelectric effect. (b) Write Einstein's photoelectric equation. (c) The work function of a metal is  $3.2 \times 10^{-19} \text{ J}$ . Calculate (i) the threshold frequency, (ii) the maximum KE of the ejected electron if light of frequency  $8 \times 10^{14} \text{ Hz}$  is incident on the metal. [PYQ 2018 AI] [5 marks]

**Marking scheme & model answers** — see companion Answer Key PDF · all PYQs traceable to actual CBSE papers · readyforboards.com · +91 70330 05444